

SYNTHESIS OF 2,24-DIEPICASTASTERONE AND 3,24-DIEPICASTASTERONE AS POTENTIAL BRASSINOSTEROID METABOLITES OF THE COCKROACH *Periplaneta americana*

Brunhilde VOIGT^{a1,*}, Andrea PORZEL^{a2}, Günter ADAM^{a3}, Dieter GOLSCH^b, Waldemar ADAM^{b1}, Christoph WAGNER^{c1} and Kurt MERZWEILER^{c2}

^a Abteilung für Naturstoffchemie, Leibniz-Institut für Pflanzenbiochemie, Weinberg 3, 06120 Halle/S., Germany; e-mail: ¹ bvoigt@ipb-halle.de, ² aporz@ipb-halle.de, ³ gadam@ipb-halle.de

^b Institut für Organische Chemie, Universität Würzburg, Am Hubland, 97074 Würzburg, Germany; e-mail: ¹ adam@chemie.uni-wuerzburg.de

^c Fachbereich Chemie, Martin-Luther-Universität, Kurt-Mothes-Str. 2, 06120 Halle/S., Germany; e-mail: ¹ c.wagner@uni-halle.de, ² merzweiler@uni-halle.de

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Investigations of the metabolic conversion of the phytohormone 24-epicastasterone (**1**) in the cockroach *Periplaneta americana* (L.) required the synthesis of 2,24-diepicastasterone (**4**), 3,24-diepicastasterone (**7b**) and 2-dehydro-3,24-diepicastasterone (**9**) as reference standards. 2,24-Diepicastasterone (**4**) was synthesized from 2 α ,3 α -epoxy derivative **2** as well as from the 2 β ,3 β -epoxy-22,23-diol **3** by acid-catalyzed water addition to the epoxy function leading to the desired 2 β ,3 α -*trans* functionality. 3,24-Diepicastasterone (**7b**) was prepared by NaBH₄-reduction of the 3-oxo derivative **6**. Upon deprotection conditions from the ketol acetonides **6** and **8** in both cases 2-dehydro-3,24-diepicastasterone (**9**) was obtained. The structure of 2,24-diepicastasterone (**4**) was confirmed by X-ray analysis.

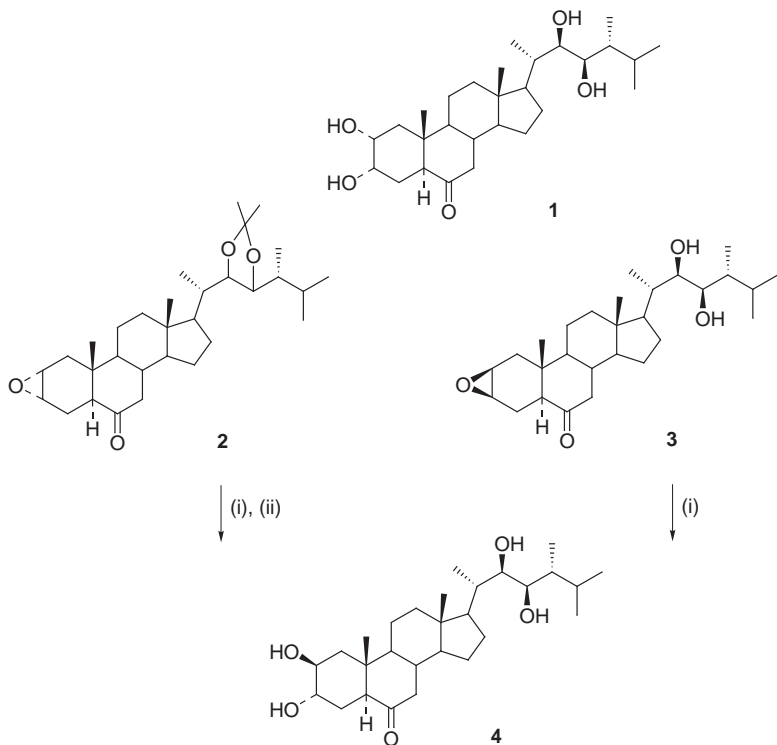
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The brassinosteroids represent a new class of steroidal phytohormones of ubiquitous occurrence in the plant kingdom with high growth-promoting and antistress activity¹. The striking structural similarity of brassinosteroids with moulting hormones of the ecdysone type² encouraged us to investigate metabolic transformations in insects. In the course of such studies, we reported recently the organ-specific epimerization of the native phytohormone 24-epicastasterone (**1**) to 2,24-diepicastasterone (**4**) in ovaries of the cockroach *Periplaneta americana* (L.), which represent the first metabolic

transformation of a brassinosteroid observed in an insect³. In this paper we present the synthesis of metabolite **4** as well as 3,24-diepicastasterone (**7b**) and 2-dehydro-3,24-diepicastasterone (**9**), required as essential reference standards in these studies.

RESULTS AND DISCUSSION

For the synthesis of 2,24-diepicastasterone (**4**) the ketal derivative of (2*R*,23*R*,24*R*)-2 α ,3 α -epoxy-22,23-dihydroxy-24-methyl-5 α -cholestan-6-one (**2**) was used, which is available in seven steps from ergosterol⁴ (Scheme 1). Hydrolytic opening of the oxirane ring in **2** with 2.5 M sulfuric acid in tetrahydrofurane-water 9 : 1 at room temperature followed by deprotection of the side chain with 4 M HCl in MeOH at 50 °C gave 2,24-diepicastasterone (**4**) in good yield, prepared also from the corresponding 2 α ,3 α -epoxy



SCHEME 1

Reagents and conditions: (i) 2.5 M H₂SO₄, THF/H₂O; (ii) 4 M HCl, MeOH, 50 °C

22,23-diol by Levinson *et al.*⁵. Acid-catalyzed epoxide opening of 2 β ,3 β -epoxy-diol⁴ **3**, detected also very recently as native phytohormone 24-epicastasterone in *Lychnis viscaria*⁶, led likewise to compound **4**. Thus, upon acid-catalyzed ring opening of both epimeric epoxides **2** and **3**, in agreement with the Fürst-Plattner-rule the same compound **4** with *trans*-diaxial 2 β ,3 α -diol function was formed. The structure of **4** was confirmed by X-ray analysis⁷ (Fig. 1), showing an intramolecular O(22)–H...O(23) hydrogen bond as well as three intermolecular hydrogen bridges to nearest neighbour molecules within the cell.

For the synthesis of 3,24-diepicastasterone (**7b**) the diisopropylidene derivative of 24-epicastasterone **5** was used as starting compound (Scheme 2). Reaction of **5** with methyl(trifluoromethyl)dioxirane⁸ (TFD) in dichloromethane during 20 h at room temperature afforded 3-dehydro-24-epicastasterone-22,23-acetonide (**6**) as main product (52%). As minor components the corresponding 2-dehydro-3,24-diepicastasterone acetonide **8** (8%), reflecting simultaneous isomerisation of **6**, as well as the 22,23-acetonide of 24-epicastasterone⁹ (10%) were obtained.

In earlier investigations we have shown the selective C-25 side-chain oxyfunctionalization of the 22,23-monoacetonide of 2,3-diacetyl-24-epicastasterone with TFD (ref.⁹). However, in the case of bisacetonide **5** the ketal function in position 2 α ,3 α is considerably more reactive towards TFD than the stronger shielded ketal in the side chain. The first step of the reaction cascade is the deprotection to the 2 α ,3 α -diol, followed by oxidation of

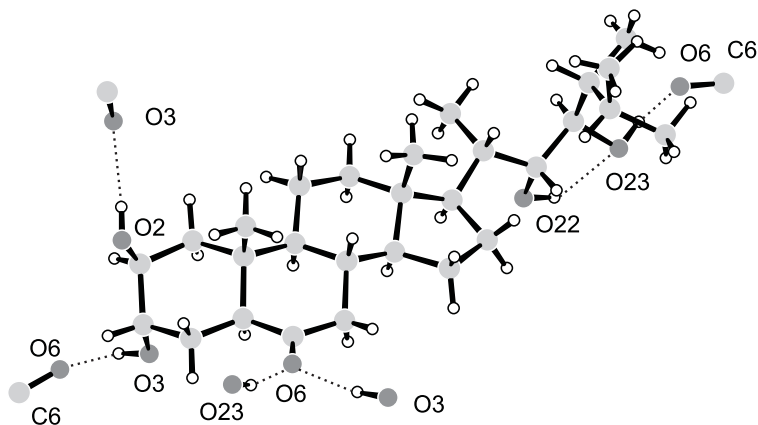
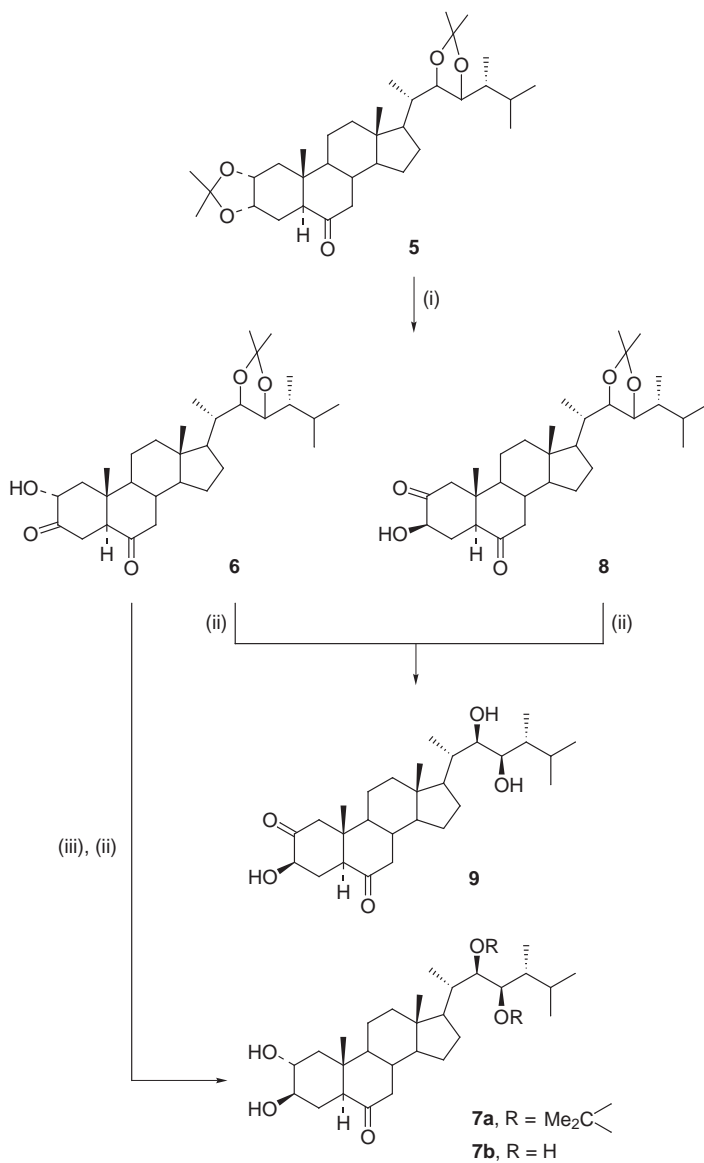


FIG. 1
Molecular structure of **4** with the hydrogen bridges



SCHEME 2

Reagents and conditions: (i) TFD, CH₂Cl₂, rt; (ii) 4 M HCl, MeOH, 50 °C; (iii) NaBH₄, EtOH, -25 °C

one of the hydroxy groups to afford mainly the 3-oxo derivative **6** besides 2-ketone **8** and the 22,23-acetonide of 24-epicastasterone. Similar results were described by Bovicelli *et al.*¹⁰ and Curci *et al.*¹¹, who used dimethyldioxirane for the monooxidation of *sec*-1,2-diols to the corresponding keto alcohols, which exploits the inhibiting effect of the carbonyl group to prevent further oxidation.

Careful reduction of compound **6** with sodium borohydride in ethanol at $-25\text{ }^{\circ}\text{C}$ furnished stereoselectively the 3β -hydroxy derivative **7a**. Recovery of the 22,23-diol function by treatment of **7a** with 4 M HCl in MeOH at $50\text{ }^{\circ}\text{C}$ led to the desired 3,24-diepicastasterone **7b**. Compound **7b**, available also from a 2α -bromo-3-oxo derivative¹², was detected as a free and acyl-conjugated metabolite of 24-epicastasterone (**1**) in cell suspension cultures of *Ornithopus sativus*¹³. Very recently 3,24-diepicastasterone was detected in immature seeds of *Phaseolus vulgaris*¹⁴. Also the 24*S*-epimer of **7b**, 3-epicastasterone, was described to be naturally occurring in *Phaseolus vulgaris* seeds¹⁵.

Deprotection of ketals of **6** and **8** with 4 M HCl in MeOH at $50\text{ }^{\circ}\text{C}$ led in both cases to the same 2-dehydro-3,24-diepicastasterone (**9**), which indicates that simultaneous isomerisation of the 2-hydroxy-3-oxo function has taken place in case of **6**. Similar rearrangements under acetic conditions to the preferred 3β -hydroxy-2-oxo compounds have been reported in the cholestane series^{16,17}. The spectral data of the new compounds are in agreement with the given structures (see Experimental). The unequivocal assignments of the ^1H and ^{13}C NMR signals were established by the combined use of one- and two-dimensional NMR experiments (COSY, HSQC, HMBC). The configuration at C-2 and/or C-3 was established by NOE difference experiments (Tables I and II).

The phytohormone activity of 2,24- and 3,24-diepicastasterone (**4** and **7b**) as well as 2-dehydro-3,24-diepicastasterone (**9**) was studied using the highly sensitive and specific rice lamina inclination assay¹⁸. The obtained results showed that the 2-epimer **4** at a concentration of 0.1 ppm has 87%, the 3-epimer **7b** 80% and the 2-dehydro derivative **9** 60% activity related to 24-epicastasterone as standard (100%). Investigations of compounds **4**, **7b** and **9** for an activity as moulting hormone showed no agonist nor antagonist properties¹⁹.

TABLE I
¹H NMR data of compounds **4**, **6**, **7a**, **7b**, **8** and **9** (in CDCl₃)

Position	δ_{H} (J, Hz)					
	4^c	6	7a	7b^c	8	9
1	1.68/1.77	1.46/2.542 dd (12.7/7.0)	1.242 dd (12.9/11.4)/2.05	1.215 dd (12.7/11.5)/2.02	2.364 d (13.3)/2.594 d (13.3)	2.361 d (13.2)/2.593 d (13.2)
2	3.828 m	4.258 ddd (12.1/7.0/3.2)	3.598 ddd (11.4/9.0/4.8)	3.525 ddd (11.5/9.1/4.9)	-	-
3	3.864 m	-	3.389 ddd (11.6/9.0/4.9)	3.314 ddd (11.7/9.1/5.0)	4.158 ddd (12.1/7.4/3.3)	4.161 dd (12.0/7.5)
4	1.63/1.98	2.518 dd (14.0/2.9)/2.705 ddd (14.0/13.6/13.4)	1.95/1.60	1.91/1.54	2.484 ddd (13.9/7.4/3.7)/1.76	2.485 ddd (13.9/7.5/3.2)/1.76
5	2.747 dd (12.4/2.3)	2.650 dd (13.4/2.7)	2.332 dd (12.6/3.0)	2.339 dd (12.6/2.9)	2.803 dd (12.7/3.1)	2.799 dd (12.7/3.2)
7	2.03/2.273 dd (13.2/4.6)	1.998 dd (13.1/12.3)/ 2.388 dd (13.1/4.4)	1.96/2.315 dd (13.2/4.6)	2.00/2.300 dd (13.3/4.6)	2.04/2.403 dd (13.4/4.5)	2.06/2.407 dd (13.4/4.5)
8	1.80	1.848 dddd (12.3/10.7/10.7/4.4)	1.784 m	1.794 m	1.75	1.76
9	1.34	1.35	1.31	1.34	1.55	1.56
11	1.65/1.35	1.69/1.44	1.65/1.34	1.64/1.36	1.53/1.38	1.54/1.38
12	1.26/2.03	1.32/2.06	1.30/2.04	1.28/2.03	1.32/2.06	1.30/2.06
14	1.33	1.32	1.31	1.32	1.32	1.33
15	1.57/1.111 m	1.58/1.11	1.58/1.097 m	1.57/1.110 m	1.59/1.116 m	1.60/1.123 m
16	1.99/1.30	2.04/1.36	2.03/1.34	2.00/1.30	2.03/1.37	2.01/1.32
17	1.57	1.55	1.54	1.57	1.55	1.60

TABLE I
(Continued)

Position	$\delta_{\text{H}}^{\text{a,b}}$ (J, Hz)								
	4 ^c	6	7a	7b ^c	8	9			
18	0.687 s	0.692 s	0.663 s	0.682 s	0.668 s	0.681 s			
19	0.955 s	1.045 s	0.804 s	0.794 s	0.713 s	0.712 s			
20	1.45	1.50	1.50	1.45	1.52	1.47			
21	0.969 d (6.7)	0.986 d (6.3)	0.981 d (6.2)	0.964 d (6.7)	0.983 d (6.3)	0.979 d (6.7)			
22	3.666 d (4.4/1.3)	3.935 d (7.0)	3.936 (br) d (7.0)	3.660 dd (4.8/1.6)	3.940 d (6.9)	3.698 dd (4.6/1.5)			
23	3.36	3.567 dd (9.4/7.0)	3.563 dd (9.4/7.0)	3.359 dd (6.0/4.8)	3.566 m	3.416 dd (6.0/4.6)			
24	1.47	1.56	1.56	1.47	1.57	1.50			
25	1.992 m	2.08	2.10	1.90	2.08	1.901 sept. d (6.8/3.8)			
26 ^{pro-R}	0.859 d (6.8)	0.813 d (6.8) ^d	0.811 d (6.8)	0.859 d (6.8)	0.812 d (6.8) ^d	0.872 d (6.8)			
27 ^{pro-S}	0.917 d (6.9)	0.911 d (7.0) ^d	0.909 d (7.0)	0.915 d (6.9)	0.911 d (7.0) ^d	0.922 d (6.9)			
28	0.833 d (7.0)	0.707 d (7.0)	0.704 d (7.0)	0.833 d (7.0)	0.707 d (7.0)	0.851 d (7.0)			
Me	-	1.342 s	1.343 s	-	1.348 s	-			
Me	-	1.387 s	1.389 s	-	1.392 s	-			

^a H- α /H- β . ^b ¹H chemical shifts without multiplet specification are chemical shifts of HSQC correlation peaks. ^c In CDCl₃ + CD₃OD.

^d Diastereotopic methyl groups 26/27 are not assigned.

TABLE II
 ^{13}C chemical shifts of compounds **4**, **6**, **7a**, **7b**, **8** and **9** (in CDCl_3)

Position	δ_{C}					
	4 ^a	6	7a	7b ^a	8	9
1	38.7	47.8	44.3	44.1	50.5	50.6
2	69.5	72.0	72.1	71.7	209.9	209.9
3	68.6	211.0	75.8	75.3	74.5	74.5
4	22.6	35.1	27.8	27.5	31.0	31.0
5	51.4	58.6	56.6	56.6	55.2	55.2
6	214.2	208.0	210.1	210.9	208.4	208.4
7	46.3	46.4	46.5	46.4	46.5	46.5
8	37.7	37.6	37.6	37.6	37.7	37.7
9	54.2	53.4	53.8	53.7	53.3	53.2
10	40.8	42.7	42.9	42.8	46.4	46.4
11	20.8	21.8	21.6	21.5	21.4	21.4
12	39.2	39.1	39.1	39.3	38.9	39.1
13	42.6	42.9	42.9	42.8	42.7	42.6
14	56.3	56.2	56.3	56.4	56.2	56.4
15	23.6	23.8	23.8	23.8	23.8	23.8
16	27.4	27.6	27.6	27.6	27.6	27.7
17	52.4	53.4	53.4	52.6	53.3	52.5
18	11.5	11.8	11.8	11.8	11.7	11.7
19	14.6	13.8	14.4	14.2	14.2	14.2
20	40.1	37.9	38.0	40.1	37.9	40.2
21	12.0	12.6	12.6	12.3	12.6	12.4
22	71.9	82.3	82.4	72.4	82.3	72.6
23	75.5	80.3	80.4	76.0	80.4	76.4
24	41.4	43.7	43.8	41.4	43.7	41.4
25	26.6	27.8	27.7	26.9	27.7	27.0
26 ^{pro-R}	16.8	16.1 ^b	16.0	17.2	15.9 ^b	17.2
27 ^{pro-S}	21.7	21.1 ^b	21.1	22.0	21.1 ^b	21.1
28	10.3	9.9	9.8	10.7	11.7	10.8
Cq		108.0	108.0		108.0	
Me		27.4	27.4		27.3	
Me		27.1	27.2		27.1	

^a In $\text{CDCl}_3 + \text{CD}_3\text{OD}$. ^b Diastereotopic methyl groups 26/27 are not assigned.

EXPERIMENTAL

General

Melting points were determined on a Boetius hot-stage microscope and are uncorrected. IR spectra (wavenumbers in cm^{-1}) were recorded on a Bruker IFS 28 instrument. Optical rotations were measured on a DIP 1000-polarimeter and are given in $10^{-1} \text{ deg cm}^2 \text{ g}^{-1}$. UV spectra were measured on a Uvikon 941 Kontron instrument. CD spectra were recorded with a Jasco J 710 spectrometer. Mass spectra (EI MS, 70 eV) were obtained with a AMD 402 spectrometer. The GC MS data of trimethylsilyl derivatives were obtained with a MD-800 Fisons instrument. The relative retention times (RR_t) values were calculated with respect to 5 α -cholestane. ^1H and 2D NMR spectra were recorded on a Varian UNITY 500 spectrometer at 499.8 MHz, whereas ^{13}C and APT spectra were determined on a Varian GEMINI 300 spectrometer at 75.5 MHz. CDCl_3 was used as solvent unless otherwise noted. TMS (δ 0) and CDCl_3 (δ 77.0) were used as internal reference for ^1H and ^{13}C spectra, respectively. Chemical shifts are given in ppm (δ -scale), coupling constants (J) in Hz. TLC plates precoated with silica gel 60 PF254 0.2 mm (Merck) and for column chromatography silica gel 60, 0.04–0.063 mm (Merck), were used. The preparative HPLC analysis was carried out on a Knauer instrument, supplied with a YMC-column, ODS, 5 mm, 20 \times 150 mm, with $\text{MeCN-H}_2\text{O}$ as eluent and UV detection at 210 nm. The elemental analyses were carried out on a LECO CHNS-932 instrument (LECO Instrumente GmbH, Kirchheim/München).

For the X-ray crystal structure determination, the data were collected on a STOE-IPDS diffractometer by using $\text{MoK}\alpha$ radiation ($\lambda = 0.71073 \text{ \AA}$) at room temperature. The structure was solved by direct methods (SHELXS86)²⁰ and all non H-atoms were refined anisotropically by full-matrix least-squares on F^2 ; H-atoms were included in calculated positions and refined as riding atoms (SHELXL93)²¹. For the graphical representations the program DIAMOND was used²².

2,24-Diepicasterone (4)

Method A. From **2**: A solution of epoxide **2** (49 mg, 0.1 mmol) in $\text{THF-H}_2\text{O}$ (9 : 1 v/v, 15 ml) was treated with 2.5 M H_2SO_4 (0.2 ml) and the mixture was stirred at room temperature for 2 h. After evaporation of the solvent the residue was extracted with CHCl_3 to give 46 mg crude product, which was heated at 50 $^\circ\text{C}$ with 4 M HCl (1 ml) in MeOH (10 ml) for 3 h. Work-up and flash chromatography by elution with ethyl acetate gave **4** (37 mg, 80%) with m.p. 234–235 $^\circ\text{C}$ and $[\alpha]_D^{25} -11.90$ (c 1.04, MeOH). HPLC: R_t 5.92, $\text{MeCN-H}_2\text{O}$ 65 : 35 v/v. IR (Nujol), ν_{max} : 3 560, 3 526, 3 454 (OH), 1 685 (CO). UV (c 1.04, MeOH), λ_{max} (ϵ): 289 (50). CD: $\Delta\epsilon_{294} -1.35$ (MeCN). EI MS, m/z (rel.%): 446 ($\text{M}^+ - 18$, 4), 393 ($\text{M}^+ - 71$, 5), 375 (393 – 18, 7), 364 ($\text{M}^+ - 100$, 100), 345 ($\text{M}^+ - 119$, 55). GC MS: RR_t 1.89. EI MS of the methylboronate-TMS-ether, m/z : 632 (M^+ , 3), 617 ($\text{M}^+ - 15$, 6), 515 (617 – 98, 67), 426 (32); HR MS, m/z : 364.2619 (calculated for $\text{C}_{22}\text{H}_{36}\text{O}_4$ 364.2624), 345.2428 (calculated for $\text{C}_{22}\text{H}_{33}\text{O}_3$ 345.2426). For $\text{C}_{28}\text{H}_{48}\text{O}_5$ calculated: 72.37% C, 10.42% H; found: 72.21% C, 10.20% H. For ^1H and ^{13}C NMR spectra see Tables I and II.

Method B. From **3**: A solution of 2 β ,3 β -epoxide **3** (23 mg, 0.05 mmol) was treated with 2.5 M H_2SO_4 as described under method A. After 5 min at room temperature the reaction was complete; work-up and crystallization from ethyl acetate–hexane gave **4** (21 mg, 86%), whose data are identical with those of **4**, synthesized as described under method A.

X-Ray Crystal Structure Determination of **4**

$C_{28}H_{48}O_5$; orthorhombic; space group $P2_12_12_1$; unit cell dimensions: $a = 6.265(2)$ Å, $b = 14.976(3)$ Å, $c = 28.035(8)$ Å, $\alpha = \beta = \gamma = 90^\circ$, $V = 2\,630.4(12)$ Å³, $Z = 4$, density (calculated) = 1.173 Mg m⁻³; absorption coefficient 0.078 mm⁻¹; $F(000) = 1\,024$. θ range: 1.99 to 26.05° ; index ranges: $-7 \leq h \leq 7$, $-18 \leq k \leq 18$, $-34 \leq l \leq 34$; reflections collected: $22\,246$; independent reflections: $5\,096$ [$R(\text{int}) = 0.1167$]; data/restraints/parameters: $5\,096/0/490$. S : 0.924 ; final R indices [$I > 2\sigma(I)$]: $R_1 = 0.0430$, $wR_2 = 0.0705$; R indices (all data): $R_1 = 0.0849$, $wR_2 = 0.0801$; absolute structure parameter: -0.0 (11); largest difference peak and hole: 0.168 and -0.143 e Å⁻³. Compound **4** has three intermolecular and one intramolecular hydrogen bridges: O(2)–H(45)···O(3) = 2.1352 ; O(3)–H(46)···O(6) = 1.9162 ; O(23)–H(48)···O(6) = 1.9300 ; O(22)–H(47)···O(23) = 2.1263 (ref.⁷).

3-Dehydro-24-epicastasterone 22,23-Acetonide (**6**) and 2-Dehydro-3,24-diepicasterone 22,23-Acetonide (**8**)

A solution of 24-epicastasterone diacetonide (**5**; 300 mg, 0.55 mmol) in CH_2Cl_2 (15 ml) was treated with a 0.2 M solution of TFD (ref.⁸) in CH_2Cl_2 (3.6 ml). After 20 h standing at room temperature the peroxide test was negative. Work-up, SiO_2 chromatography and elution with hexane–ethyl acetate 75 : 25 v/v gave the 3-oxo derivative **6** as main product (144 mg, 52%), m.p. 189 – 192 °C and $[\alpha]_D^{25} +4.32$ (c 1.27, MeOH). IR (Nujol), ν_{max} : 3 500 (OH), 1 721 (CO). UV (c 1.27, MeOH), λ_{max} (ϵ): 290 (105), 256 (175). EI MS, m/z : 502 (M^+ , 2), 487 ($M^+ - 15$, 88), 431 ($M^+ - 71$, 28), 387 (431 – 44, 24), 301 (32), 171 (96), 142 (100); HR MS, m/z : 487.3442 (calculated for $C_{30}H_{47}O_5$ 487.3461), 171.1391 (calculated for $C_{10}H_{19}O_2$ 171.1397), 142.1356 (calculated for $C_9H_{18}O$ 142.1355). For $C_{31}H_{50}O_5$ calculated: 74.06% C, 10.02% H; found: 74.12% C, 10.03% H. For ¹H and ¹³C NMR spectra see Tables I and II.

Further elution with hexane–ethyl acetate 6 : 4 v/v furnished the 2-dehydro derivative **8** (22 mg, 8%) with m.p. 121 – 124 °C and $[\alpha]_D^{25} +2.10$ (c 1.01, MeOH). IR (Nujol), ν_{max} : 3 478 (OH), 1 704 (CO). UV (c 1.01, MeOH), λ_{max} (ϵ): 259 (270). CD: $\Delta\epsilon_{311} -0.65$ (MeOH); $\Delta\epsilon_{303} -0.56$; $\Delta\epsilon_{276} +0.72$; $\Delta\epsilon_{234} +0.22$; $\Delta\epsilon_{202} -1.08$. EI MS, m/z : 503 ($M^+ + 1$, 23), 487 ($M^+ - 15$, 78), 431 ($M^+ - 71$, 29), 387 (431 – 44, 27), 301 (38), 171 (100), 142 (87), 99 (64). HR MS, m/z : 487.3438 (calculated for $C_{30}H_{47}O_5$ 487.3452), 431.2813 (calculated for $C_{26}H_{39}O_5$ 431.2828), 301.1819 (calculated for $C_{19}H_{25}O_3$ 301.1835), 171.1380 (calculated for $C_{10}H_{19}O_2$ 171.1375), 142.1349 (calculated for $C_9H_{18}O$ 142.1341), 99.0809 (calculated for $C_6H_{11}O$ 99.0808). ¹H and ¹³C NMR spectra see Tables I and II.

3,24-Diepicasterone 22,23-Acetonide (**7a**)

A solution of **6** (20 mg, 0.04 mmol) in dry EtOH (10 ml) and $NaBH_4$ (2 mg, 1 equivalent) was stirred under argon at -25 °C for 5 min. TLC monitoring showed two new more polar products, which were separated by preparative HPLC. The fraction eluted with MeCN– H_2O 95 : 5 had R_f 8.95 and was the acetonide **7a** (17 mg, 85%) with m.p. 228 – 231 °C (ref.¹² gives m.p. 235 – 236 °C) and $[\alpha]_D^{25} -51.4$ (c 1.02, MeOH). UV (c 1.27, MeOH), λ_{max} (ϵ): 280 (483). CD: $\Delta\epsilon_{299} -0.84$ (MeCN). EI MS, m/z : 489 ($M^+ - 15$, 38), 433 ($M^+ - 71$, 15), 389 (489 – 100, 17), 301 (18), 171 (69), 142 (100), 99 (80). For ¹H and ¹³C NMR spectra see Tables I and II.

The fraction with R_f 9.58 was identical with the known 22,23-acetonide of 24-epicastasterone (ref.⁹).

3,24-Diepicasterone (7b)

The acetonide **7a** (12 mg) was deprotected by stirring of the methanolic solution (6 ml) with 4 M HCl (0.6 ml) at 50 °C for 1 h. Work-up and crystallization (CHCl₃) gave the desired 3,24-diepicasterone (**7b**; 9 mg, 80%) with m.p. 209–212 °C (ref. gives m.p. 213–215 °C) and $[\alpha]_D^{23}$ -52.16 (c 0.533, MeOH). HPLC: R_t 5.47 (MeCN–H₂O 65 : 35 v/v). UV (c 0.53, MeOH), λ_{\max} (ε): 285 (195). CD: $\Delta\epsilon_{298}$ -0.86 (MeCN). EI MS, m/z : 364 (M⁺ - 100, 100), 345 (45), 319 (38). GC MS: RR_t = 2.04. EI MS of the methylboronate–TMS–ether, m/z : 512 (M⁺ - 120, 8), 358 (9), 287 (10), 155 (100). HR MS, m/z : 364.2597 (calculated for C₂₂H₃₆O₄ 364.2580), 363.2520 (calculated for C₂₂H₃₅O₄ 363.2505), 362.2453 (calculated for C₂₂H₃₄O₄ 362.2449), 346.2479 (calculated for C₂₂H₃₄O₃ 346.2450), 345.2400 (calculated for C₂₂H₃₃O₃ 345.2370), 319.2262 (calculated for C₂₀H₃₁O₃ 319.2251). For ¹H and ¹³C NMR spectra see Tables I and II.

2-Dehydro-3,24-diepicasterone (9)

Method A. From **6**: Acetonide **6** (27 mg, 0.05 mmol) in MeOH (5 ml) was stirred with 4 M HCl (0.5 ml) at 50 °C for 2 h. Work-up and SiO₂ chromatography gave **9** (20 mg, 80%) with m.p. 125–128 °C and $[\alpha]_D^{26}$ + 2.50 (c 0.80, MeOH). UV (c 0.08, MeOH), λ_{\max} (ε): 275 (578). CD: $\Delta\epsilon_{299}$ -0.04 (MeOH); $\Delta\epsilon_{234}$ +0.02. EI MS, m/z : 462 (M⁺, 4), 458 (12), 391 (M⁺ - 71, 8), 362 (M⁺ - 100, 100), 361 (M⁺ - 101, 99), 343 (361 - 18, 86). HR MS, m/z : 362.2419 (calculated for C₂₂H₃₄O₄ 362.2381), 361.2357 (calculated for C₂₂H₃₃O₄ 361.2335), 343.2255 (calculated for C₂₂H₃₁O₃ 343.2236), 332.2339 (calculated for C₂₁H₃₂O₃ 332.2327), 101.0981 (calculated for C₆H₁₃O 101.0996). For ¹H and ¹³C NMR spectra see Tables I and II.

Method B. From **8**: Acetonide **8** (25 mg, 0.05 mmol) was deprotected in the same manner as described in method A. After SiO₂ chromatography, a product was separated (15 mg, 60%), whose physical data were identical with that of **9**, described under method A and was synthesized from **6**.

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REFERENCES AND NOTES

1. For reviews, see: a) Adam G., Porzel A., Schmidt J., Schneider B., Voigt B. in: *Studies in Natural Products Chemistry* (Atta-ur-Rahman, Ed.), Vol. 18, p. 495. Elsevier, Amsterdam–London–New York–Tokyo 1996; b) Khrupach V. A., Zhabiniskii V. N., de Groot A. E.: *Brassinosteroids – a New Class of Plant Hormones*. Academic Press, San Diego 1999; c) Sakurai A., Yokota T., Clouse S. D. (Eds): *Brassinosteroids, Steroidal Plant Hormones*. Springer, Tokyo 1999; d) Adam G., Schmidt J., Schneider B.: *Prog. Chem. Org. Nat. Prod.* **1999**, 78, 1.
2. Lafont R.: *Arch. Insect Biochem. Physiol.* **1997**, 35, 3.
3. Schmidt J., Richter K., Voigt B., Adam G.: *Z. Naturforsch., C* **2000**, 55, 233.
4. Voigt B., Takatsuto S., Yokota T., Adam G.: *J. Chem. Soc., Perkin Trans. 1* **1995**, 1495.
5. Levinson E. E., Kuznetsova N. A., Podkhalyuzina N. Y., Traven Y. F.: *Mendeleev Commun.* **1994**, 96.

6. Friebe A., Schmidt J., Volz A., Voigt B., Adam G., Schnabl H.: *Phytochemistry* **1999**, *52*, 1609.
7. Crystallographic data for the structure reported in this paper have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication number CCDC-102822. Copies of the data can be obtained free of charge on application to CCDC, e-mail: deposit@ccdc.cam.ac.uk.
8. Adam W., Smerz A. K.: *Bull. Soc. Chim. Belg., Eur. Sect.* **1996**, *105*, 581.
9. Voigt B., Porzel A., Golsch D., Adam W., Adam G.: *Tetrahedron* **1996**, *52*, 10653.
10. Bovicelli P., Lupattelli P., Sanetti A.: *Tetrahedron Lett.* **1995**, *36*, 30.
11. Curci R., D'Accolti L., Dinoi A., Fusco C., Rosa A.: *Tetrahedron Lett.* **1996**, *37*, 115.
12. Levinson E. E., Traven V. F.: *J. Chem. Res. (S)* **1996**, 196.
13. Kolbe A., Schneider B., Porzel A., Adam G.: *Phytochemistry* **1996**, *41*, 163.
14. Park S. C., Kim T.-W., Kim S.-K.: *Bull. Korean Chem. Soc.* **2000**, *21*, 1274.
15. Yokota T., Kim S. K., Ogino Y., Takahashi N. in: *Various Brassinosteroids from Phaseolus vulgaris Seeds: Structures and Biological Activity* (A. R. Cooke, Ed.). Presented at the *Proc. 14th Annu. Plant Growth Regulator Soc. America Meeting Honolulu 1987*, p. 28.
16. Williamson K. L., Johnson W. S.: *J. Org. Chem.* **1961**, *26*, 4563.
17. Henbest H. B., Jones D. N., Slater G. P.: *J. Chem. Soc.* **1961**, 4472.
18. Arima M., Yokota T., Takahashi N.: *Phytochemistry* **1984**, *23*, 1587.
19. Voigt B., Whiting P., Dinan L.: *Cell. Mol. Life Sci.* **2001**, *58*, 1133.
20. *Programs for Crystal Structure Determination*. University of Göttingen, Göttingen 1986.
21. *Programs for Crystal Structure Determination*. University of Göttingen, Göttingen 1993.
22. Brandenburg K.: *Informationssystem für Kristallstrukturen, Version 1.0.3*. Crystal impact gbR, Bonn 1996.